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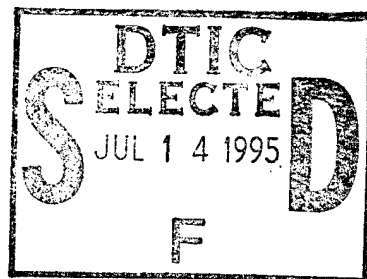


Point-Charge Analysis of Symmetry-  
Preserving Charge Compensation and  
Vacancies in the Fluorapatites  
 $\text{Ca}_5(\text{PO}_4)_3\text{F}$  and  $\text{Sr}_5(\text{PO}_4)_3\text{F}$

by Clyde A. Morrison

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13. ABSTRACT (Maximum 200 words)  A single point-charge model was used to investigate four types of charge compensation and two types of vacancies for the M2 site in $\text{M}_5(\text{PO}_4)_3\text{F}$ ( $\text{M} = \text{Ca}, \text{Sr}$ ). Two effective charges were chosen for the oxygen in the $(\text{PO}_4)^{3-}$ complex. A set of crystal-field components, $A_{nm}$ , was obtained for each assumed compensation or vacancy, and the crystal-field parameters, $B_{nm}$ , for $\text{Nd}^{3+}$ were calculated for each type. The resulting energy levels for $\text{Nd}^{3+}$ were then calculated and the results for each type compared, showing a rather complicated result for the $[(L,S)]$ multiplets higher in energy than the $^4F_{3/2}$ multiplet of $\text{Nd}^{3+}$ . Unsuccessful attempts were made to perform a simple correlation analysis, such as the rotational invariants of the crystal-field parameters and the width of the energy splittings of the $^4I_j$ and $^4F_{3/2}$ multiplets of $\text{Nd}^{3+}$ for each case considered. A calculation for the M1 site is included, and the energy splittings of the $^4I_j$ and $^4F_{3/2}$ are shown to be comparable to the energy splittings for the M2 sites.				
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# 1. Introduction

The fluorapatites have received considerable attention recently as hosts for rare-earth ions,  $Ln^{3+}$ , and transition-metal ions in various ionization states [1–5]. In the fluorapatites  $M_5(BO_4)_3F$ , the  $M$  ion can be  $Ca^{2+}$  or  $Sr^{2+}$ , which have ionic radii matching  $Ln^{3+}$ ; lanthanide ions can replace the  $M^{2+}$  ion [6]. (For simplicity, the various fluorapatites  $M_5(BO_4)_3F$  can be referred to as MBAP.) However, the replacement of the  $M^{2+}$  ion by  $Ln^{3+}$  must be accompanied by charge compensation, achieved either by the inclusion of some optically neutral impurity ion, such as  $Na^+$  ( $2Ca^{2+} \rightarrow Na^+ + Ln^{3+}$ ) or  $Si^{4+}$  ( $Ca^{2+} + P^{5+} \rightarrow Si^{4+} + Ln^{3+}$ ), or by vacancy creation ( $3Ca^{2+} \rightarrow 2Ln^{3+} +$  a vacant  $Ca^{2+}$  site). Other more sophisticated forms of charge compensation can be and have been suggested [7–9]. In the analysis reported here, I concentrate on two types of charge compensation, which were proposed by Maximova and coworkers some time ago [6,9]. I also consider the case where the charge compensation is remote and unidentified, and the case where the neodymium ion enters the M2 site and a calcium vacancy occurs at one of the two nearest M2 sites in the crystal.

This report considers the fluorapatites  $Ca_5(PO_4)_3F$  (CPAP) and  $Sr_5(PO_4)_3F$  (SPAP), referred to collectively as MPAP. In these fluorapatites, the crystal belongs to the  $P6_3/m$  (176) space group. The  $M$  ions occupy two sites: M1 in the  $4f$  site with  $C_3$  symmetry and M2 in the  $6h$  site with  $C_s$  symmetry [10]. The latter site, M2, is the subject here, and I assume that the  $Nd^{3+}$  ions occupy this site, which is the site that is generally assumed to be occupied by the rare-earth ions in the apatite hosts.

Three of the six M2 sites (in the unit cell) of the  $P6_3/m$  space group are (in hexagonal crystal coordinates)  $(x, y, 1/4)$ ,  $(-y, x - y, 1/4)$ , and  $(y - x, -x, 1/4)$ , and the three others are given by  $(x, y, z) \rightarrow (-x, -y, -z)$ ; the crystal-field components are always computed at the  $(x, y, 1/4)$  site. The group  $C_s$  has only one operation, which is a reflection in a plane perpendicular to the principal axis, which, in this particular case, is the  $c$ -axis of the hexagonal crystal; since I am interested only in the symmetry-preserving charge compensators, I restrict my attention to the first set ( $z = 1/4$ ) given above. To preserve the  $C_s$  symmetry, I consider the charge compensator ions occupying the sites with  $z = 1/4$ , and in all cases I take the origin, the location of the  $Ln^{3+}$  ion, at  $(x, y, 1/4)$ . Thus, where we have oxygen compensation, I assume that the fluorine ion at  $(0, 0, 1/4)$  is replaced by an oxygen ion. For sodium compensation, I assume that the calcium ions at either  $(-y, x - y, 1/4)$  or  $(y - x, -x, 1/4)$  are replaced by sodium ions (or any ion with +1 values that might occupy the site). Both these types of charge compensation are discussed in some detail by Maksimova and Sobol' [7]. The detailed x-ray data on MPAP for  $M = Ca$  and  $Sr$  are given in table 1 for the representative ions in each site, and the coordinates (hexagonal) for the equivalent sites of each ion are given in the International Tables [10].

Table 1. Crystallographic data on  $M_5(BO_4)_3F$  ( $M = Ca, Sr; B = P, V$ ).

Hexagonal  $C_{6h}^2$  ( $P6_3/m$ ) 176,  $Z = 2$

Ion	Site	Symmetry	$x$	$y$	$z$	$q^a$
M1	4f	$C_3$	2/3	1/3	$z$	2
M2	6h	$C_s$	$x$	$y$	1/4	2
B	6h	$C_s$	$x$	$y$	1/4	$q_B$
O1	6h	$C_s$	$x$	$y$	1/4	$q_O$
O2	6h	$C_s$	$x$	$y$	1/4	$q_O$
O3	12i	$C_1$	$x$	$y$	$z$	$q_O$
F	2a	$C_{3h}$	0	0	1/4	-1

X-ray data on  $M_5(BO_4)_3F$

M	B	$a$ (Å)	$c$ (Å)	$Z_{M1}$	$X_{M2}$	$Y_{M2}$	$X_B$	$Y_B$	$X_{O1}$	$Y_{O1}$
Ca <sup>b</sup>	P	9.3973	6.872	0.0010	0.00712	0.24827	0.36895	0.3985	0.4849	0.3272
Sr <sup>c</sup>	V	10.0077	7.4342	0.0004	0.24943	0.01059	0.3682	0.3982	0.4706	0.5962
Sr <sup>c</sup>	P	9.7153	7.2810	-0.0002	0.23910	-0.01445	0.3992	0.3685	0.3306	0.4810

M	B	$X_{O2}$	$Y_{O2}$	$X_{O3}$	$Y_{O3}$	$Z_{O3}$
Ca <sup>b</sup>	P	0.4667	0.5875	0.2575	0.3421	0.0705
Sr <sup>c</sup>	V	0.4835	0.3160	0.2498	0.3403	0.0671
Sr <sup>c</sup>	P	0.5356	0.1183	0.3441	0.2612	0.0788

<sup>a</sup>Two values of  $q_O$  have been chosen, -2 and -1.8, with  $q_P = -3 - 4q_O$  in each case.

<sup>b</sup>J. M. Hughes, M. Cameron, and K. D. Crowley, *Structural Variations in Natural F, OH, Cl Apatites*, Am. Mineral. **74** (1989), 870.

<sup>c</sup>Private communication, R. E. Peale, University of Central Florida.

## 2. Theory

In all the calculations reported here, the crystal-field Hamiltonian for the  $4f^N$  configuration is given by

$$H_{CEF} = \sum_{n \text{ even}} \sum_{m=-n}^n B_{nm}^* \sum_{i=1}^N C_{nm}(i), \quad (1)$$

where the  $B_{nm}$  are the crystal-field parameters and the  $C_{nm}$  are

$$C_{nm}(i) = \sqrt{4\pi/(2n+1)} Y_{nm}(\theta_i, \varphi_i), \quad (2)$$

with

$$C_{n-m} = (-1)^m C_{nm}^*$$

and the  $Y_{nm}$  are the ordinary spherical harmonics. In  $C_s$  symmetry, the values of  $m$  are restricted by  $n + m = 0, \pm 2, \pm 4$ , with  $|m| \leq n$  and  $n = 2, 4$ , and 6. The parameters entering into the free-ion Hamiltonian are from aqueous solution [11], and their values as well as the details of the computation are given elsewhere [12,13]. The crystal-field splitting is predominantly determined by the  $n$ -even  $B_{nm}$ , and, from the discussion above, the number of

$B_{nm}$  for  $C_s$  symmetry is limited to 14, with  $B_{22}$  real and positive; these properties of  $B_{22}$  are simply achieved by a rotation about the z-axis (c-axis). In the three-parameter crystal-field theory [14], the crystal-field parameters,  $B_{nm}$ , are related to the crystal-field components,  $A_{nm}$ , by

$$B_{nm} = \rho_n A_{nm} , \quad (3)$$

where the  $\rho_n$  values represent the effective values of  $\langle r^n \rangle$  for the rare-earth ion. Values of  $\rho_n$  have been tabulated for the entire rare-earth series [12]. The crystal-field components are given by

$$A_{nm} = -e^2 \sum_j q_j C_{nm}(\hat{R}_j) / R_j^{n+1} , \quad (4)$$

where  $q_j$  is the effective charge in units of electronic charge on the ion at  $R_j$  [15], and the sum is over all the ions in the solid. Equation (4) has been used to calculate the  $A_{nm}$  for the Ca2 site in CPAP and the Sr2 site in SPAP with the x-ray data reported in table 1. Two values of the oxygen charge have been used:  $q_O = -2$  (for which  $q_{Ca} = 2$  and  $q_P = 5$ ) and  $q_O = -1.8$  (for which  $q_{Ca} = 2$  and  $q_P = -4.2$ ), which in both cases maintain the total charge of the  $PO_4$  molecule at  $-3$  ( $q_P = -3 - 4q_O$ ). The value of  $-1.8$  for the oxygen charge was chosen because this value best fit the experimental value of the point charge  $A_{nm}$  determined experimentally for rare-earth ions in  $Y_3Al_5O_{12}$  (yttrium aluminum garnet—YAG). Table 2 has the results of the calculations of the  $A_{nm}$  for the  $C_s$  sites at  $(x, y, 1/4)$  in CPAP and SPAP with no charge compensation. The  $A_{nm}$  for CPAP given in table 2 should correspond to the conjecture of Maksimova and Sobol' [7] for low concentration of  $Nd^{3+}$  ions.

To calculate the effect of replacing the nearest  $F^-$  site at  $(0, 0, 1/4)$  by an  $O^{2-}$  site, I calculate the  $A_{nm}$  for a negative charge at this site and add these  $A_{nm}$  to the corresponding values given in table 2.

Table 2. Even- $n$  crystal-field components,  $A_{nm}$  ( $cm^{-1}/\text{\AA}^n$ ), for  $M2$  site at  $(x, y, 1/4)$  in CPAP and SPAP (no charge compensation).<sup>a</sup>

$A_{nm}$	CPAP <sup>b</sup>				SPAP <sup>c</sup>			
	$q_O = -1.8$		$q_O = -2$		$q_O = -1.8$		$q_O = -2$	
	Re	Im	Re	Im	Re	Im	Re	Im
$A_{20}$	9481	—	11398	—	6535	—	8081	—
$A_{22}$	-2048	-506.5	-2384	-717.6	527.2	-1664	483.1	-2034
$A_{40}$	2031	—	2012	—	1902	-1911	—	—
$A_{42}$	2829	-1801	3187	-1983	-1815	1053	-2021	1204
$A_{44}$	589.8	-2164	369.5	-2417	1124	-1298	1386	-1251
$A_{60}$	90.85	—	121.2	—	113.2	—	141.5	—
$A_{62}$	-329.9	-284.8	-355.9	-315.5	-3.465	-321.1	-7.119	-349.6
$A_{64}$	156.0	70.42	193.1	79.35	-113.7	-39.64	-135.9	-55.88
$A_{66}$	64.57	-158.4	85.17	-175.0	47.50	179.5	63.08	197.7

<sup>a</sup> $x$  and  $y$  for Ca2 and Sr2 sites are given in table 1.

<sup>b</sup>Smallest Ca2 – F distance = 2.3109 Å, smallest Ca2 – Ca2 distance = 4.0025 Å (two ions).

<sup>c</sup>Smallest Sr2 – F distance = 2.3962 Å, smallest Sr2 – Sr2 distance = 4.1504 Å (two ions).

Consider for a moment CPAP: the nearest  $F^-$  to the Ca2 site is at 2.3109 Å, and the next nearest  $F^-$  ions are at 4.1434 Å; these facts would lead one to surmise that most of the contribution of all the fluorine ions is approximately due to this nearest ion. That this is the case is shown in table 3 for CPAP; if a rough estimate of the effect of the nearest  $F^-$  being replaced by an  $O^{2-}$  ion is needed, one can obtain it by adding in the contribution of all the  $F^-$  ions to the total. That is, simply double the contribution of the  $F^-$  ions to the crystal-field components  $A_{nm}$ . However, I do not make that approximation here but evaluate the crystal-field components for a single negative charge at (0, 0, 1/4) and add this to the total  $A_{nm}$  for the site at (x, y, 1/4).

Table 3. Crystal-field components for a single  $F^-$  ion at (0, 0, 1/4) and total  $F^-$  ion contribution to crystal-field components at Ca2 site at (x, y, 1/4) in  $Ca_5(PO_4)_3F$  (CPAP).<sup>a</sup>

$A_{nm}$	Single $F^-$		All $F^-$ ions	
	Re	Im	Re	Im
$A_{11}$	-386	15374	-421	17646
$A_{20}$	-4706	—	-2972	—
$A_{22}$	-5756	-289	-6454	-318
$A_{31}$	44.2	-1763	31.3	-1247
$A_{33}$	171	-2270	179	-2327
$A_{40}$	661	—	644	—
$A_{42}$	696	34.9	604	30.3
$A_{44}$	917	92.3	934	92.9
$A_{51}$	-6.55	261	-6.86	274
$A_{53}$	-21.2	281	-20.3	266
$A_{55}$	-47.3	375	-47.5	375
$A_{60}$	-103	—	-108	—
$A_{62}$	-106	-5.30	-110	-5.54
$A_{64}$	-115	-11.6	-112	-11.4
$A_{66}$	-155	-23.5	-155	-23.5
$A_{71}$	1.05	-41.7	1.06	-42.3
$A_{73}$	3.26	-43.3	3.35	-44.6
$A_{75}$	6.00	-47.6	5.95	-47.2
$A_{77}$	11.4	-64.4	11.4	-64.4

<sup>a</sup>All  $A_{nm}$  values in units of  $cm^{-1}/\text{\AA}^n$ .

### 3. Calculation of Crystal-Field Parameters, $B_{nm}$ , for Various Types of Charge Compensation

The types of charge compensation considered are given in table 4, and the labels given in table 4 that identify the type of charge compensation are used hereafter. As an example of the method of obtaining crystal-field parameters for  $Nd^{3+}$ , I give the details of the calculation for the C1 type of charge compensation in table 4. The results given in table 2 are for a crystal where each ion in the solid is in the position given in table 1. Then to account for an  $Na^+$  ion replacing a  $Ca^{2+}$  ion at  $(-y, x - y, 1/4)$ , we can calculate the  $A_{nm}$  for a single charge of -1 at that point and add these  $A_{nm}$  to the appropriate  $A_{nm}$  of table 2. After this, the resulting  $A_{nm}$  are rotated so that



$A_{22}$  is real and positive [16]. Using the rotated  $A_{nm}$ , we can calculate the crystal-field parameters for  $\text{Nd}^{3+}$  by using equation (3). For the D types of table 4, we can create a vacancy at the two  $\text{Ca}^{2+}$  sites by choosing a  $-2$  charge at the D1 or D2 site in the table, and the resulting  $A_{nm}$  are treated as in the C1 case discussed above. The results for all the cases of table 4 are given in tables 5 and 6. The invariants,  $I_n(B)$ , given by [17]

$$I_n(B) = \left[ \sum_{m=-n}^n B_{nm}^* B_{nm} \right]^{1/2}, \quad (5)$$

were calculated and are also given in tables 5 and 6. The results given in tables 5 and 6 show that the crystal-field parameters,  $B_{nm}$ , for the different types of charge compensation vary considerably; except for the B type ( $\text{F}^- \rightarrow \text{O}^{2-}$ ), however, these variations have little effect on the  $I_n(B)$  for the different forms of charge compensation. The result found here for six of the possibly numerous charge compensation possibilities is indicative of the

**Table 4. Types of charge compensation considered in text.<sup>a</sup>**

Type	Remark
A	No charge compensation
B	$\text{Ca}^{2+}$ and $\text{F}^-$ are replaced by $\text{O}^{2-}$ at $(0, 0, 1/4)$ , and $\text{Nd}^{3+}$ at $(x, y, 1/4)$
C1	$\text{Ca}^{2+}$ at $(-y, x - y, 1/4)$ replaced by $\text{Na}^+$
C2	$\text{Ca}^{2+}$ at $(y - x, -x, 1/4)$ replaced by $\text{Na}^+$
D1	$\text{Ca}^{2+}$ at $(-y, x - y, 1/4)$ removed (a charge $-2$ at this site)
D2	$\text{Ca}^{2+}$ at $(y - x, -x, 1/4)$ removed (a charge $-2$ at this site)

<sup>a</sup>Crystal field is calculated for the site at  $(x, y, 1/4)$ , with  $x$  and  $y$  for the  $\text{Ca}^{2+}$  site given in table 1. In all cases read  $\text{Ca}^{2+}$  as meaning  $\text{Ca}^{2+}$  or  $\text{Sr}^{2+}$ . ( $\text{Na}^+$  can be taken as any possible replacement ion with valence of  $+1$ .)

**Table 5.  $\text{Nd}^{3+}$  crystal-field parameters,  $B_{nm}$  ( $\text{cm}^{-1}$ ), and rotation invariants,  $I_n(B)$  ( $\text{cm}^{-1}$ ), for different types of charge compensation in CPAP with  $q_O = -1.8$  and  $-2$ .<sup>a</sup>**

$B_{nm}$	$q_O = -1.8$						$q_O = -2$					
	A	B	C1	C2	D1	D2	A	B	C1	C2	D1	D2
$B_{20}$	1617	815	1463	1463	1309	1309	1944	1142	1790	1790	1636	1636
$B_{22}$	360	1338	458	505	672	601	425	1399	511	572	642	739
$B_{40}$	1173	1555	1197	1197	1222	1222	1162	1544	1186	1186	1211	1211
$\text{Re}B_{42}$	-1337	-1922	-1795	-907	-611	-1951	-1432	-2087	-1933	-1025	-2141	-778
$\text{Im}B_{42}$	1402	1221	788	1709	1816	360	1627	1392	1032	1908	564	2020
$\text{Re}B_{44}$	-281	611	711	-908	-1106	1152	-593	394	401	-1104	996	-1266
$\text{Im}B_{44}$	-1264	-1348	-1115	-879	-541	-701	-1282	-1483	-1384	-831	-1081	-475
$B_{60}$	144	-19.5	141	141	137	137	193	28.7	189	189	186	186
$\text{Re}B_{62}$	618	735	448	684	699	316	686	791	531	746	393	761
$\text{Im}B_{62}$	314	389	527	128	24.3	616	317	416	537	146	645	44.5
$\text{Re}B_{64}$	272	82.4	201	217	156	92.1	326	146	286	258	175	188
$\text{Im}B_{64}$	-16.5	78.5	188	-165	-225	262	-64.3	74.3	172	-211	288	-277
$\text{Re}B_{66}$	90.7	224	-211	253	274	-274	127	217	-197	281	-308	312
$\text{Im}B_{66}$	256	233	173	104	-13.2	-10.9	282	254	242	133	49.3	14.1
$I_2(B)$	1696	2061	1600	1628	1617	1560	2035	2285	1930	1964	1871	1941
$I_4(B)$	3498	4143	3552	3480	3464	3606	3839	4437	3894	3822	3950	3807
$I_6(B)$	1130	1272	1131	1135	1139	1132	1262	1369	1263	1267	1264	1272

<sup>a</sup>See table 4 for explanation of symbols used for different types of charge compensation.

Table 6.  $\text{Nd}^{3+}$  crystal-field parameters  $B_{nm}$  ( $\text{cm}^{-1}$ ), and rotation invariants,  $I_n(B)$  ( $\text{cm}^{-1}$ ) for different types of charge compensation in SPAP with  $q_O = -1.8$  and  $-2$ .<sup>a</sup>

$B_{nm}$	$q_O = -1.8$						$q_O = -2$					
	A	B	C1	C2	D1	D2	A	B	C1	C2	D1	D2
$B_{20}$	1115	395	976	976	838	838	1379	659	1240	1240	1102	1102
$B_{22}$	298	1178	422	397	570	533	357	1234	485	443	633	567
$B_{40}$	1099	1417	1119	1119	1139	1139	1104	1422	1124	1124	1144	1144
$\text{Re}B_{42}$	-896	-1299	-602	-1160	-436	-1241	-946	-1404	-672	-1249	-508	-1367
$\text{Im}B_{42}$	-816	-745	-1049	-413	-1127	-160	-975	-857	-1179	-582	-1257	-295
$\text{Re}B_{44}$	-99.2	515	-672	645	-846	932	-390	333	-839	387	-971	813
$\text{Im}B_{44}$	987	1015	693	783	408	457	1005	1120	640	1033	341	781
$B_{60}$	180	52.8	177	177	175	175	225	97.7	222	222	219	220
$\text{Re}B_{62}$	485	600	513	386	507	305	538	644	558	453	554	370
$\text{Im}B_{62}$	-159	-197	4.26	-335	89.7	-410	-139	-204	6.52	-323	88.5	-416
$\text{Re}B_{64}$	184	29.8	177	89.3	140	5.55	233	77.8	209	160	166	65.8
$\text{Im}B_{64}$	-52.6	-91.7	73.8	-172	131	-197	-17.8	-94.6	106	-173	165	-230
$\text{Re}B_{66}$	113	232	289	-218	283	-294	176	243	322	-176	322	-315
$\text{Im}B_{66}$	-273	-276	-60.3	-199	84.2	-18.2	-279	-294	-71.0	-279	72.0	-99.6
$I_2(B)$	1192	1712	1144	1126	1162	1127	1469	1865	1417	1389	1419	1363
$I_4(B)$	2472	2355	2458	2519	2446	2565	2690	3188	2678	2738	2665	2786
$I_6(B)$	895	735	897	896	899	897	997	1115	1000	999	1002	1000

<sup>a</sup>See table 4 for explanation of symbols used for different types of charge compensation.

number of lines that appear in the absorption and emission spectra, as was for example observed by Zounani et al [18] for  $\text{Eu}^{3+}$  in SPAP.

The crystal-field parameters given in tables 5 and 6 were used in calculating the energy levels for  $\text{Nd}^{3+}$ . The details of the procedure used in the calculation are given elsewhere [19]. The calculation covered the lowest 13 multiplets, with a total of 60 energy levels. Instead of presenting all the results, which are quite extensive, I have chosen to give only the width in energy of each multiplet where such a quantity can be reasonably determined. The results for all the  $B_{nm}$  given in tables 5 and 6 are presented in tables 7 and 8. The widths of the  $^4I_J$  ( $J = 9/2$  through  $15/2$ ) and the  $^4F_{3/2}$  are well determined, and greater significance should be given to these multiplets than to the widths of the higher multiplets when one compares these values to experiment. Because of the very large magnitude of the  $B_{nm}$ , in particular  $B_{2m}$ , the higher multiplets are severely mixed. In a number of cases, the mixture was so great that the number of levels of a multiplet allowed by group theory was exceeded. This occurred numerous times for the  $^4S_{3/2}$  levels, which had three levels identified (the levels are identified by the largest coefficient of the state in the wave function for that state), where by group theory there should be only two. This result does not indicate an error in the calculation; it simply means that the crystal field is so strong that in these spectral regions it renders meaningless the method used to identify the energy levels by assignment to particular multiplets in the weak crystal-field limit. Since the crystal-field parameters given in tables 5 and 6 are reasonable approximations to the values that would be obtained by fitting the spectra, these results indicate the difficulties to be expected in the analysis of experimental data.

Table 7. Total crystal-field splittings,  $\Delta E_J$  ( $\text{cm}^{-1}$ ), of each  $J$  multiplet of  $\text{Nd}^{3+}$  in CPAP in  $C_s$  site with  $q_O = -1.8$  and  $-2$ .<sup>a</sup>

[(L,S)]	$q_O = -1.8$						$q_O = -2$					
	A	B	C1	C2	D1	D2	A	B	C1	C2	D1	D2
$^4I_{9/2}$	649	747	632	645	645	624	732	795	714	729	702	729
$^4I_{11/2}$	506	608	482	504	503	460	581	632	557	578	535	577
$^4I_{13/2}$	589	714	564	586	584	542	676	746	652	673	630	671
$^4I_{15/2}$	849	994	823	850	849	800	980	1071	955	980	931	981
$^4F_{3/2}$	228	394	235	238	258	254	262	439	270	276	289	298
$^4S_{3/2}^b$	29	66	32	32	37	38	32	70	37	35	42	40
$^4F_{9/2}$	449	584	440	450	459	458	513	638	500	515	517	525
$^2H_{11/2}$	114	147	105	109	107	103	136	150	125	130	119	127
$^2G_{7/2}^c$	306	402	319	300	294	331	344	431	359	336	373	329

<sup>a</sup>The  $^4F_{5/2}$ ,  $^2H_{9/2}$ , and  $^4F_{7/2}$  are intermixed, as are  $^4G_{5/2}$  and  $^2G_{7/2}$ .

<sup>b</sup>The  $^4S_{3/2}$  multiplet is very strongly mixed with the  $^4F_{7/2}$  multiplet.

<sup>c</sup>The  $^2G_{7/2}$  multiplet is strongly mixed with the  $^4G_{5/2}$  multiplet, and levels belonging to  $^2G_{7/2}$  were arbitrarily selected.

Table 8. Total crystal-field splittings,  $\Delta E_J$  ( $\text{cm}^{-1}$ ), of each  $J$  multiplet of  $\text{Nd}^{3+}$  in SPAP in  $C_s$  site with  $q_O = -1.8$  and  $-2$ .<sup>a</sup>

[(L,S)]	$q_O = -1.8$						$q_O = -2$					
	A	B	C1	C2	D1	D2	A	B	C1	C2	D1	D2
$^4I_{9/2}$	464	586	461	451	462	449	527	619	525	512	526	505
$^4I_{11/2}$	362	493	360	340	360	321	422	511	420	400	420	381
$^4I_{13/2}$	427	580	424	405	424	385	497	605	495	475	493	455
$^4I_{15/2}$	623	807	623	600	624	580	727	1859	727	705	727	683
$^4F_{3/2}$	173	309	178	178	192	195	207	347	215	211	231	226
$^4S_{3/2}^b$	36	49	36	37	26	27	28	53	29	30	30	33
$^4F_{9/2}$	314	457	318	317	332	336	364	495	367	362	380	378
$^2H_{11/2}$	81	125	78	77	77	79	96	128	93	90	92	90
$^2G_{7/2}^c$	235	369	234	246	234	258	229	393	228	265	228	250

<sup>a</sup>The  $^4F_{5/2}$ ,  $^2H_{9/2}$ , and  $^4F_{7/2}$  are intermixed, as are  $^4G_{5/2}$  and  $^2G_{7/2}$ .

<sup>b</sup>The  $^4S_{3/2}$  multiplet is very strongly mixed with the  $^4F_{7/2}$  multiplet.

<sup>c</sup>The  $^2G_{7/2}$  multiplet is strongly mixed with the  $^4G_{5/2}$  multiplet, and levels belonging to  $^2G_{7/2}$  were arbitrarily selected.

As a final computation, the crystal-field components,  $A_{nm}$ , were calculated for the M1 site with  $C_3$  symmetry in CPAP and SPAP, and the crystal-field parameters were calculated for  $\text{Nd}^{3+}$ . In these calculations, charge compensation was ignored. The results for both CPAP and SPAP are given in table 9 along with the invariants  $I_n(B)$  for each case. This calculation corresponds to type A in table 4 for the M2 site. On comparing  $I_n(B)$  from table 5 for type A with the  $I_n(B)$  of table 9 for CPAP, we find that  $I_2(B)$  and  $I_4(B)$  for  $C_s$  symmetry are larger than  $I_2(B)$  and  $I_4(B)$  for  $C_3$  symmetry. However, the opposite is true for the corresponding values of  $I_6(B)$ , where  $I_6(B)$  for the M1 site in CPAP is larger than  $I_6(B)$  for the M2 site in CPAP. This same result is true for the  $I_n(B)$  in SPAP. The energy widths of the multiplets are given in table 10, and a comparison of the results for case A in table 7 shows mixed results; that is, some of the widths of the multiplets for the

Table 9.  $\text{Nd}^{3+}$  crystal-field parameters,  $B_{nm}$  ( $\text{cm}^{-1}$ ), and rotational invariants,  $I_n(B)$  ( $\text{cm}^{-1}$ ), for  $C_3$  (M1) site in CPAP and SPAP with  $q_O = -1.8$  and  $-2$ .

$B_{nm}, I_n(B)$	CPAP		SPAP	
	$q_O = -1.8$	$q_O = -2$	$q_O = -1.8$	$q_O = -2$
$B_{20}$	1247	1556	936	1173
$B_{40}$	-2252	-2359	-1585	-1646
$B_{43}$	1016	1114	700	759
$B_{60}$	-1176	-1312	-814	-909
$\text{Re}B_{63}$	890	935	629	673
$\text{Im}B_{63}$	-260	-432	65.7	194
$\text{Re}B_{66}$	-117	-3.61	-130	-35.9
$\text{Im}B_{66}$	356	417	-249	-311
$I_2(B)$	1247	1556	936	1173
$I_4(B)$	2671	2837	1869	1965
$I_6(B)$	1544	2047	1064	1415

Table 10. Total crystal-field splitting,  $\Delta E_J$  ( $\text{cm}^{-1}$ ), of each  $J$  multiplet of  $\text{Nd}^{3+}$  in CPAP and SPAP in  $C_3$  (M1) site with  $q_O = -1.8$  and  $-2$ .

$[(L,S)]$	CPAP		SPAP	
	$q_O = -1.8$	$q_O = -2$	$q_O = -1.8$	$q_O = -2$
$^4I_{9/2}$	667	735	471	529
$^4I_{11/2}$	373	426	260	298
$^4I_{13/2}$	439	4892	308	347
$^4I_{15/2}$	861	944	595	652
$^4F_{3/2}$	144	167	117	139
$^4S_{3/2}$	26	30	17	21
$^4F_{9/2}$	367	422	268	308
$^2H_{11/2}$	142	165	98	114
$^2G_{7/2}$	324	362	233	255

M1 sites in table 10 are larger than the corresponding multiplets for the M2 sites given in table 7. The results for SPAP given in table 10 and for case A of table 8 are similar to the result for CPAP.

## 4. Discussion and Conclusion

A number of the calculations presented here were made in an attempt to find some general trend and eventually a simple criterion that could be used to sort out the various sites in the experimental data. The rotational invariants are one such quantity that I chose in the attempt to correlate the  $I_n(B)$  with the  $\Delta E_J$  of various multiplets. In particular, if we select the  $^4F_{3/2}$  splittings and compare them with the  $I_2(B)$  for the different charge compensations, the results for the splittings  $\Delta E$  ( $^4F_{3/2}$ ) are  $B > D1 > D2 > C2 > C1 > A$ ; for  $I_2(B)$ , on the other hand,  $B > A > C2 > D1 > C1 > D2$ . If the crystal field were weaker, one would expect a one-to-one correlation between  $I_2(B)$  and  $\Delta E$  ( $^4F_{3/2}$ ), since for weak crystal fields,  $\Delta E$  ( $^4F_{3/2}$ ) must be proportional to  $I_2(B)$  when  $J$ -mixing by the crystal field is ignored. However, that  $J$ -mixing cannot be ignored is obvious from the above results. An examination of the composition of the  $^4F_{3/2}$  eigenfunctions reveals that the  $^4F_{3/2}$  lowest energy level wavefunction for case B in CPAP ( $q_O = -2$ ) is approximately

$$90\% {}^4F_{3/2} + 6\% {}^4F_{5/2} + 1\% {}^4G_{5/2};$$

for the same case, the highest level for the  ${}^4F_{3/2}$  eigenfunction is

$$91\% {}^4F_{3/2} + 3\% {}^4F_{5/2} + 3\% {}^4F_{7/2}.$$

For case A, the wavefunctions for the corresponding levels are

$$95\% {}^4F_{3/2} + 3\% {}^4F_{5/2} + 0.50\% {}^4F_{7/2}$$

for the lowest energy, and

$$92\% {}^4F_{3/2} + 5\% {}^4F_{5/2} + 1\% {}^4F_{7/2}$$

for the highest. This admixture of the wavefunctions of different  $J$  values, caused by the strength of the crystal field, makes a simple first analysis of the experimental observation of the splitting of the  ${}^4F_{3/2}$  impossible. Obviously, for the higher energy multiplets, which are not so well isolated from one another, the mixing is confusingly severe: For case A in CPAP ( $q_O = -2$ ), the wavefunction for the lower  ${}^4S_{3/2}$  is

$$71\% {}^4S_{3/2} + 24\% {}^4F_{7/2} + 2\% {}^4F_{5/2}$$

and for the highest is

$$83\% {}^4S_{3/2} + 15\% {}^4F_{7/2} + 1\% {}^4F_{9/2}.$$

Thus, the operator equivalent method or any other method that ignores  $J$ -mixing by the crystal field would fall far short of the mark of a good representation of the experimental situation.

The relationship of the  ${}^4I_J \Delta E_J$  to the  $I_n(B)$  is quite complicated even in theory, and one might analyze a direct relation to the  $B_{nm}$  themselves by using some of the results given by Karayianis [20] where  $J$ -mixing by the crystal field is included.

Finally, a calculation of the crystal-field parameters for  $\text{Nd}^{3+}$  in the M1 site (table 1) with  $C_3$  symmetry for  $q_O = -1.8$  and  $-2$  was performed for the case of no charge compensation; these results are given in table 9 for CPAP and SPAP. This calculation corresponds to the A type of calculation of table 4. The  $I_n(B)$  values in comparison with the corresponding values of  $I_n(B)$  for the M1 in table 9 show that  $I_2(B)$  and  $I_4(B)$  are larger for the M2 site than for the M1 site, but the reverse is true for the corresponding values of  $I_6(B)$ . However, the  $\Delta E_J$  for the different multiplets of the  ${}^4I$  term show erratic behavior; that is, for the  $q_O = -1.8$  in CPAP, the  $\Delta E_{9/2}$  for M2 are smaller than those for M1; for  $\Delta E_{11/2}$ ,  $M2 > M1$ ; for  $\Delta E_{13/2}$ ,  $M2 > M1$ ; for  $\Delta E_{15/2}$ ,  $M2 < M1$ . Similar results are obtained for SPAP. Thus it is apparently not possible to separate the experimental data by sites by identifying the width of the splitting of even well-defined multiplets ( ${}^4I_J$ ), unless a careful analysis of these splittings using some of the results of Karayianis [20] revealed a somewhat simple relationship of these splittings of the multiplets with the crystal-field parameters.

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